



J-1612040701040400 Seat No. _____

**Master of Pharmacy Management (Sem. IV)
(CBCS) Examination**

June / July - 2019

Instrumental Analysis-II

Time : 3 Hours]

[Total Marks : 80

nstructions :

- (1) Answer and tie up both sections separately.
- (2) Figure to the right indicates marks.
- (3) Answer the **three (3)** questions from each section.
- (4) Question one (1) and question **five (5)** are compulsory.
- (5) Draw neat and clean diagrams as required.

SECTION – I

- 1** Answer any **seven** out of **10** : **14**
- (a) What is pulse polarography?
 - (b) Define equivalence conductance.
 - (c) Define optical rotation and specific rotation.
 - (d) Define diffusion current, migration current and residual current.
 - (e) Differentiate Normal phase chromatography & Reverse Phase chromatography
 - (f) What is resolution ? Give acceptance criteria for resolution.
 - (g) What is signal to noise ratio?
 - (h) What is retention time and elution time?
 - (i) Comment on : If HETP value is low, the efficiency of the column is Higher.
 - (j) How multiple extraction is benefited to single extraction?
- 2** (a) Explain various techniques for development of Paper **7**
in Paper chromatography.
- (b) Explain Plate and Rate theory of chromatographic **6**
separation.

- 3 (a) What is thermogravimetry? Write a short note on TGA. 7
 (b) Write a short note on pH metrology. 6
- 4 (a) Write a short note on : Conductometric titrations. 7
 (b) Write a short note on polarimetry. 6

SECTION – II

- 5 Answer two out of three : 14
 (a) Explain the principle, methodology and application of thin layer chromatography.
 (b) Write a short note on DME and Derive the Ilkovic equation.
 (c) State and explain Kohlrausch Law. Give its applications. Describe in brief factors affecting electrolytic conductance.
- 6 (a) Enlist characteristics of an ideal reference electrode used in potentiometry. Give a brief account of various reference electrodes. 7
 (b) What is the difference between DTA and DSC? State the applications of both the above methods. 6
- 7 (a) Write short note on oxygen combustion flask method. 7
 (b) What is polarography? Explain the basic principle of Polarography. Explain modification of polarography. 6
- 8 (a) What is extraction? Discuss the factors affecting solvent extraction. 7
 (b) Partition coefficient is 4 in ether/H₂O system, compare the efficiency of extraction of 10 ml aqueous solution of compound with,
 a) 40 ml portion of ether
 b) 2 times 20 ml portion of ether
 c) 4 times 10 ml portion of ether.
 Give the comment 6